Contents lists available at ScienceDirect



Bioorganic & Medicinal Chemistry

journal homepage: www.elsevier.com/locate/bmc

2'-Deoxy-1-methylpseudocytidine, a stable analog of 2'-deoxy-5-methylisocytidine

Hyo-Joong Kim[†], Nicole A. Leal[‡], Steven A. Benner^{*}

Westheimer Institute for Science and Technology, Foundation for Applied Molecular Evolution, PO Box 13174, Gainesville, FL 32604, United States

ARTICLE INFO

Article history: Received 29 December 2008 Revised 20 March 2009 Accepted 22 March 2009 Available online 27 March 2009

Keywords: iso-Cytidine Pseudocytidine Synthetic biology Expanded genetic alphabets Clinical diagnostics

ABSTRACT

2'-Deoxy-5-methylisocytidine is widely used in assays to personalize the care of patients infected with HIV, hepatitis C, and other infectious agents. However, oligonucleotides that incorporate 2'-deoxy-5methylisocytidine are expensive, because of its intrinsic chemical instability. We report here a C-glycoside analog that is more stable and, in oligonucleotides, pairs with 2'-deoxyisoguanosine, contributing to duplex stability about as much as a standard 2'-deoxycytidine and 2'-deoxyguanosine pair. © 2009 Elsevier Ltd. All rights reserved.

1. Introduction

One success story arising from efforts to create artificially expanded genetic information systems (AEGIS) come from the use of 2'-deoxy-5-methylisocytidine (isoC, 1, Fig. 1) and 2'-deoxyisoguanosine (isoG, 2, Fig. 1) as fifth and sixth 'letters' in the DNA genetic alphabet.^{1,2} The isoC presents a hydrogen bond 'acceptor-acceptor-donor' hydrogen bonding pattern on a small pyrimidine; isoG presents a hydrogen bond 'donor-donor-acceptor' hydrogen bonding pattern on a large purine nucleobase. Therefore, isoC and isoG can form a Watson-Crick pair having the same geometry as the T:A and C:G base pairs, but are joined by a differ-



Figure 1. The chemical structures of 2'-deoxy/socytidine (1), 2'-deoxy/soguanosine (2) and 2'-deoxy-5-methylpseudocytidine (3).

E-mail addresses: hkim@ffame.org (H.-J. Kim), nleal@ffame.org (N.A. Leal), sbenner@ffame.org (S.A. Benner).

[†] Tel.: +1 386 418 8085; fax: +1 352 271 7076.

¹ Tel.: +1 352 375 8680; fax: +1 352 271 7076.

0968-0896/\$ - see front matter © 2009 Elsevier Ltd. All rights reserved. doi:10.1016/j.bmc.2009.03.047

ent hydrogen bonding scheme. This gives DNA molecules containing *iso*C and *iso*G the property known as 'orthogonality'; they pair with each other without pairing with DNA built only from natural dA, T, dC, and dG. At the same time, the close structural similarity of the nonstandard and standard pairs means that polymerases and other enzymes used as standard tools in molecular biology can also be used with this synthetic genetic system.

Orthogonality enabled by a fifth and sixth letters of a genetic alphabet can, in principle, support low-noise microarrays, multiplexed PCR, and low cost re-sequencing of the human genome, among other applications. In commerce, the orthogonality provided by the *isoC*:*isoG* pair underlies the high sensitivity and broad dynamic range of the branched DNA Versant diagnostics systems that received FDA approval in 2003 and 2004.³ Developed by scientists at Chiron and Bayer and now sold by Siemens Medical Solutions, these permit the very sensitive detection of the nucleic acids from infectious agents in complex biological media. These tools personalize the care of some 400,000 patients infected with the HIV, hepatitis B, and hepatitis C viruses.⁴

The *isoC*:*isoG* pair has also been used as a fifth and sixth letter in a PCR-amplifiable artificial genetic system.^{5,6} This represents one of the first artificial chemical systems capable of Darwinian evolution, meeting some of the criteria for artificial life.

As one of its disadvantages, however, 2'-deoxy-5-methylisocytidine undergoes facile cleavage of the glycosidic bond under acidic conditions. This propensity for acid-catalyzed depyrimidinylation can be used as a tool to detect *iso*C in oligonucleotides⁵ and is managed during phosphoramidite synthesis by careful handling of formamidine protected phosphoramidite precursors. Nevertheless,

^{*} Corresponding author. Tel.: +1 352 271 7005; fax: +1 352 271 7076.

these requirements increase the cost of oligonucleotides that contain 2'-deoxy-5-methylisocytidine, especially in human diagnostics where regulatory standards are rigorous. An analog of *iso*C that still pairs with 2'-deoxyisoguanosine but does not suffer depyrimidinylation is desirable.

One analog that might have these properties is 2'-deoxy-1methylpseudocytidine (ψ C, **3**, Fig. 1), the C-glycoside analog of 2'deoxy-5-methylisocytidine. Further, a methyl group on the pyrimidine nitrogen should prevent tautomerization and epimerization, undesired reactions seen with other AEGIS components.⁷ However, given the known propensity of C-glycosides to change the conformation of the 2'-deoxyribose ring, we were concerned that incorporating a significant number of C-glycosides would significantly alter the conformation of a DNA duplex that is made by this strand.

Very recently, we reported experiments that showed that the conformation of a duplex is not altered even if twelve C-glycosides are incorporated sequentially.⁸ Further, we showed that certain polymerases could incorporate five consecutive C-glycosides in template-directed synthesis from triphosphates. This provided the motivation to examine the analogous C-glycoside of *iso*-cytidine.

We report here the synthesis of this species, a new compound, in protected form, as well as the synthesis of a test set of oligonucleotides. We report that when incorporated into an oligonucleotide at the appropriate position, 2'-deoxy-1-methylpseudocytidine forms nucleobase pairs with *iso*G that contribute to duplex stability about as well as the natural C:G nucleobase pair. Further, discrimination against mismatches is at the same level as seen with standard nucleobases. Therefore, this compound is a costeffective replacement for a 2'-deoxyisocytidine that will support the synthesis and use of AEGIS-containing oligonucleotides at lower cost.

2. Results and discussion

Pseudothymidine (7), prepared from pseudouridine,⁹ is potentially a precursor for protected 2'-deoxy-1-methylpseudocytidine (9). As pseudouridine is expensive and the N(1)-methylation of pseudouridine requires 5 days at reflux, an alternative route was sought. Coupling of glycal (4) and 5-iodo-1-methyluracil (5) via a palladium-catalyzed Heck reaction, followed by desilylation and diastereoselective reduction, gave pseudothymidine (7) in 51% yield (Fig. 2), a result consistent with results obtained similar palladium-mediated glycal-aglycone couplings.¹⁰

Pseudothymidine having both sugar oxygens protected (8) was converted to the correspondingly protected 2'-deoxy-1-methylpseudocytidine (9) by treatment first with 2,4,6-triisopropylbenzenesulfonyl chloride, DMAP and triethylamine in CH_3CN , and then with ammonium hydroxide (75% yield). The structure of 9 was established by 1 H and 13 C NMR and high resolution mass spectrometry.

With the 2'-deoxy-1-methylpseudocytidine 9 in hand, the phosphoramidite derivative 13 suitable for oligonucleotide synthesis was prepared as shown in Figure 3. Compound 9 was treated with benzoyl chloride in pyridine/ CH_2Cl_2 to give dibenzoylated compound 10. Treatment of 10 with excess triethylamine trihydrofluoride in THF at 45 °C for 2 days gave 11 in 85% yield. The phosphoramidite derivative 13 for DNA synthesis was synthesized via dimethoxytritylation of 11 and subsequent phosphitylation of 12 using a standard procedure.

Oligonucleotides were constructed to permit melting studies with both perfectly matched duplexes and duplexes containing mismatches. The standard sequence was identical to that used by Geyer et al.¹¹ and by their predecessor, Horn et al.¹² This allowed direct comparison of the results obtained here with their earlier results. Melting temperatures are illustrated in Table 1.

In a variety of contexts, the *isoC*:*isoG* base pair contributes more to duplex stability than any of the other standard nucleobase pairs. The reason for this is unclear, but the *isoC*:*isoG* pair appears to be the strongest of any joined by three hydrogen bonds.

At least in the context in the two test oligonucleotides, the pair between 2'-deoxy-1-methylpseudocytidine and 2'-deoxyisoguanosine contributes to duplex stability like the standard dC:dG. This result is consistent with other results obtained by Geyer et al.,¹¹ who found that an oligonucleotide duplex having a C-glycoside had a melting temperature 1–2 degrees lower than an analogous duplex with an N-glycoside.

To demonstrate the improved stability of oligonucleotides containing 2'-deoxy-1-methylpseudocytidine above that displayed by analogous oligonucleotides containing 2'-deoxyisocytidine, four oligonucleotides were prepared (Table 2) and purified by PAGE.

Treatment of the oligonucleotides with gamma-labeled ATP generated 5'-kinased oligonucleotides. These oligonucleotides were treated in parallel experiments with (a) heat at 95 °C for various times (Fig. 4) and (b) 0.05 M acetic acid followed by 0.1 M ammonium hydroxide (Fig. 5).

Treatment of oligonucleotides containing 2'-deoxyisocytidine (oligo 1 and 2 in Fig. 4) with heat at 95 °C gave fragmentation of the oligonucleotides and the intensity of fragments were increased with prolonged treatment time. However, treatment of oligonucleotides containing 2'-deoxy-1-methylpseudocytidine instead of *iso*cytidine (oligo 4 and 5 in Fig. 4) with heat at 95 °C gave no cleavage up to 90 min treatment. This result showed 2'-deoxy-1-methylpseudocytidine is more stable than 2'-deoxyisocytidine under thermal condition at neutral pH.



Figure 2. Synthesis of protected 2'-deoxypseudocytidine (9). Conditions: (a) Pd(OAc)₂, Ph₃As, Bu₃N, DMF, 60 °C, 18 h; (b) TBAF, HOAc; (c) NaBH(OAc)₃, HOAc, CH₃CN, 0 °C, 2 h (51% for three steps); (d) TBDMSCl, imidazole, DMF (93%); (e) 2,4,6-triisopropylbenzenesulfonyl chloride, DMAP, TEA, CH₃CN, rt, 24 h; (f) NH₄OH, rt, 4 h (75% for two steps).



Figure 3. Synthesis of 2'-deoxypseudocytidine phosphoramidite (13). Conditions: (a) BzCl, pyridine (74%); (b) Et₃N-3HF, THF, 45 °C, 2 days (86%); (c) DMTrCl, pyridine (66%); (d) 2-cyanoethyl diisopropylchlorophosphoramidite, DIPEA, CH₂Cl₂ (78%).

 Table 1

 Melting temperatures (°C) for nucleobase pairs

Matched nucleobase pairs Mismatched nucleobase pairs					
N1-N2	Measured	Geyer (2003) ¹¹	N ₁ -N ₂ Me	asured Gey	er (2003) ¹¹
iG-iC	62.5	63.3	iG-C 52.	5 52.	6
íG-ψC G-C	60.2 59.6	 59.5	iG-A 42. iG-T 51.		
ic-ig	59.8 61.3	61.5	iG-G 53.		
ψC–iG	59				
C-G	60.1	58.5	lege oppige		ti ang

The sequence of the DNA duplex used in the thermal denaturation experiments are 5^{2} -CACN₁ACTTTCTCTCT- 3^{2}

3'-TG T G N2 T G A A AG A G G-5'

Table 2

Oligonucleotides used in thermal and acidic stability study

Oligomer no. Sequence	Comment
1 5'-GGAGAAAGT-IC-GTGT-3'	iC at position 10
	iC at position 4
人名英格兰德 人名英格兰 机动物 机杂制 海上的过去式和过去分词 医马克克尔氏白色的 化合物出现分析 化分析出现 化合物出现 化合物出现 化合物分子 第一次问题 机油	ΨC at position 10
5 5'-CAC-ΨC-ACITICTCCT-3'	ΨC at position 4

The acid-stability of oligonucleotides containing 2'-deoxyisocyti- dine and 2'-deoxy-1-methylpseudocytidine was investigated by the incubation of the oligonucleotides in 0.05 M acetic acid at 90 °C for 0-90 min. Treatment of oligonucleotides containing 2'deoxyisocytidine (oligo 1 and 2 in Fig. 5) in 0.05 M acetic acid gave fragmentation of the oligonucleotides and complete disappearance of the starting 14-mer after 60 min. The acid treatment of oligonucleotides containing 2'-deoxy-1-methylpseudocytidine instead of iso-cytidine (oligo 4 and 5 in Fig. 5) in 0.05 M acetic acid gave fragmentation of the oligonucleotides but even after 90 min, small fraction of the starting 14-mer of oligo 4 and 5 was not cleaved. This result showed 2'-deoxy-1-methylpseudocytidine is more resistant to acidic depyrimidinylation.

3. Conclusion

2'-Deoxy-1-methylpseudocytidine (1-deaza-5-aza-5-methylisocytidine) is more robust chemically than the 2'-deoxy-5methylisocytidine that is widely used in assays to personalize the care of patients infected with HIV, hepatitis, and other infectious agents, assay for genetic disease, and support advanced architectures for human genome sequencing and analysis. In one respect,



Figure 4. Thermostability of oligonucleotides containing 2'-deoxyisocytidine and 2'-deoxy-5-methylpseudocytidine. Samples (0.3 pmol) were resolved on a 20% PAGE.

it joins with the 6-aza-2'-deoxyisocytidine, which is also reported to display acid stability.¹³ Further, it supports duplex pairing well with complementary nucleotides as part of an expanded genetic information system. Oligonucleotides that incorporate it are considerably less expensive to prepare in high quality than oligonucleotides that incorporate 2'-deoxy-5-methylisocytidine. Therefore, we expect 2'-deoxy-1-methylpseudocytidine to become an important new addition to synthetic genetic systems, especially in practical application.

4. Experimental

4.1. Pseudothymidine (7)

A mixture of Pd(OAc)₂ (36 mg, 0.16 mmol) and AsPh₃ (98 mg, 0.32 mmol) in anhydrous DMF (10 mL) was stirred at rt for



Figure 5. Acid-catalyzed hydrolysis of oligonucleotides containing 2'-deoxyisocytidine and 2'-deoxy-5-methylpseudocytidine. Starting 14-mers moved in electrophoresis at slightly different rates.

30 min. The solution was then added to a mixture of glycal 4 (280 mg, 0.80 mmol), 5-iodo-1-methyluracil (5) (200 mg, 0.80 mmol), and tributylamine (0.28 mL, 0.12 mmol) in DMF (10 mL). The resulting mixture was stirred at 60 °C for 18 h. After being cooled, the mixture was treated with HOAc (0.2 mL) and TBAF (1 M in THF, 2 mL) and stirred at rt for 1 h. Volatiles were removed by rotary evaporation under reduced pressure. The product was purified by flash chromatography (silica, gradient CH₂Cl₂/ MeOH = 15:1 to 10:1). The appropriate fraction (by TLC) was collected, evaporated and dissolved in acetic acid/acetonitrile (7 mL/ 7 mL). To this solution were added NaBH(OAc)₃ (370 mg, 1.75 mmol) at 0 °C. The mixture was stirred for 2 h, volatiles were removed by rotary evaporation under reduced pressure, and the residue was resolved by flash chromatography (silica, gradient $CH_2Cl_2/MeOH = 7:1$ to 4:1) to give 7 as a white solid (100 mg, 51%). TLC (CH₂Cl₂/MeOH, 5:1): Rf 0.25.

4.2. 3',5'-Di-O-tert-butyldimethylsilyl-pseudothymidine (8)

To a mixture pseudothymidine (7) (100 mg, 0.41 mmol) and imidazole (113 mg, 1.65 mmol) in anhydrous DMF (4 mL) was added TBDMSCI (154 mg, 1.03 mmol). The mixture was stirred at rt overnight, volatiles were removed by rotary evaporation under reduced pressure, and the product was isolated by flash chromatography (silica, gradient ethyl acetate/hexane = 1:2 to 1:1) to give **8** as a colorless viscous liquid (180 mg, 93%). TLC (EtOAc/hexane, 1:1): R_f 0.6. ¹H NMR (300 MHz, CDCl₃) δ 8.43 (br s, 1H), 7.32 (d, 1H, J = 1.2 Hz), 5.02 (m, 1H), 4.32 (m, 1H), 3.87 (m, 1H), 3.60–3.72 (m, 2H), 3.34 (s, 3H), 2.31 (ddd, 1H, J = 12.8, 6.0, 2.4 Hz), 1.70–1.82 (m, 1H), 0.90 (s, 9H), 0.89 (s, 9H), 0.07 (s, 6H), 0.06 (s, 6H).

4.3. 2'-Deoxy-3',5'-di-O-*tert*-butyldimethylsilyl-1methylpseudocytidine (9)

To a mixture of 3',5'-di-O-*tert*-butyldimethylsilyl-pseudothymidine (8) (200 mg, 0.42 mmol), 2,4,6-triisopropylbenzenesulfonyl chloride (257 mg, 0.85 mmol) and DMAP (51 mg, 0.42 mmol) in CH₃CN (15 mL) was added triethylamine (0.24 mL, 1.70 mmol) at rt and stirred for 24 h. The mixture was treated with saturated NH₄OH (7 mL) and stirred at rt for 4 h. Volatiles were removed by rotary evaporation under reduced pressure and the residue was dissolved in CH₂Cl₂ (40 mL) and washed with brine. The organic layer was collected, dried and evaporated. The product was purified from the residue by flash chromatography (silica, EtOAc/hexane = 1:1 to CH₂Cl₂/MeOH = 20:1) to give 9 as a light yellow solid (150 mg, 75%). TLC (CH₂Cl₂/MeOH, 20:1): $R_{\rm f}$ 0.15. ¹H NMR (300 MHz, CDCl₃) δ 7.20 (s, 1H), 4.82 (dd, 1H, *J* = 11.1, 5.4 Hz), 4.39 (d, 1H, *J* = 6.6 Hz), 3.7–4.0 (m, 3H), 3.41 (s, 3H), 2.2–2.3 (m, 1H), 1.75–1.85 (m, 1H), 0.90 (s, 18H), 0.08 (m, 12H). ¹³C NMR (75 MHz, CDCl₃) δ 163.99, 144.87, 103.93, 88.71, 77.84, 74.04, 63.51, 41.27, 37.65, 26.09, 26.00, 25.23, 18.74, 18.23, -4.37, -4.49, -5.28, -5.31. HRMS calcd for C₂₂H₄₃N₃O₄Si₂ + H⁺: 470.2865, found: 470.2907.

4.4. 2'-Deoxy-3',5'-di-O-*tert*-butyldimethylsilyl-1-methyl-*N*,*N*-dibenzoyl-pseudocytidine (10)

A mixture of 2'-deoxy-3',5'-di-O-*tert*-butyldimethylsilyl-1methylpseudocytidine (9) (75 mg, 0.16 mmol), benzoyl chloride (74 µL, 0.64 mmol) and anhydrous pyridine (0.5 mL) in CH₂Cl₂ (5 mL) was stirred at rt overnight. Volatiles were removed by rotary evaporation under reduced pressure. The product was purified by flash chromatography (silica, ethyl acetate/hexane = 1:5 to 1:3) to give **10** as a white solid (80 mg, 74%). TLC (EtOAc/hexane, 1:3): $R_f 0.3$. ¹H NMR (300 MHz, CDCl₃) δ 8.28 (m, 2H), 8.12 (m, 2H), 7.36-7.64 (m, 7H), 5.35 (dd, 1H, *J* = 10.2, 5.4 Hz), 4.38 (m, 1H), 3.98 (m, 1H), 3.60–3.80 (m, 2H), 3.43 (s, 3H), 2.59 (m, 1H), 1.70 (m, 1H), 0.97 (s, 9H), 0.91 (s, 9H), 0.14 (s, 3H), 0.12 (s, 3H), 0.10 (s, 6H). ¹³C NMR (75 MHz, CDCl₃) δ 179.44, 158.85, 148.94, 141.75, 137.34, 133.90, 132.63, 130.41, 130.10, 128.69, 128.33, 116.91, 87.82, 74.51, 74.16, 63.87, 43.22, 37.07, 26.14, 26.02, 18.56, 18.24, -4.40, -4.47, -5.11, -5.13.

4.5. 2'-Deoxy-1-methyl-N-benzoyl-pseudocytidine (11)

A mixture of 2'-deoxy-3',5'-di-O-tert-butyldimethylsilyl-1methyl-*N*,*N*'-dibenzoyl-pseudocytidine (**10**) (310 mg, 0.46 mmol) and triethylamine trihydrofluoride (0.6 mL) in THF (15 mL) was stirred at 45 °C for two days. Volatiles were removed by rotary evaporation under reduced pressure, and the product was purified by flash chromatography (silica, CH₂Cl₂/MeOH = 15:1 to 12:1) to give **11** as a white solid (137 mg, 86%). TLC (CH₂Cl₂/MeOH, 12:1): R_f 0.3. ¹H NMR (300 MHz, CDCl₃/CD₃OD) δ 8.16 (d, 2H, *J* = 7.8 Hz), 7.70 (s, 1H), 7.3-7.5 (m, 3H), 5.24 (m, 1H), 4.21 (m, 1H), 3.84 (m, 1H), 3.5-3.7 (m, 2H), 3.37 (s, 3H), 2.4-2.6 (m, 1H), 1.8-1.9 (m, 1H).

4.6. 2'-Deoxy-5'-dimethoxytrityl-1-methyl-N-benzoylpseudocytidine (12)

A mixture of 2'-deoxy-1-methyl-N-benzoyl-pseudocytidine (11) (129 mg, 0.37 mmol) and 4,4'-dimethoxytritylchloride (128 mg, 0.38 mmol) in anhydrous pyridine (6 mL) was stirred at rt overnight. Volatiles were removed by rotary evaporation under reduced pressure. The product was purified by flash chromatography (silica, EtOAc/hexane = 1:2 to 1:1 then CH₂Cl₂/MeOH = 20:1) to give 12 as a white solid (160 mg, 66%). TLC (CH₂Cl₂/MeOH, 20:1): R_f 0.2. ¹H NMR (300 MHz, CDCl₃) δ 8.2–8.3 (m, 2H), 7.61 (d, 1H, J = 1.5 Hz), 7.2–7.6 (m, 12H), 6.8–6.9 (m, 4H), 5.39 (t, 1H, J = 7.5 Hz), 4.45 (m, 1H), 4.06 (m, 1H), 3.80 (s, 6H), 3.35 (m, 1H), 3.17 (s, 3H), 2.6–2.7 (m, 1H), 2.0–2.2 (m, 1H). ¹³C NMR (75 MHz, CDCl₃) δ 158.82, 158.73, 148.78, 144.88, 142.36, 137.22, 135.95, 132.59, 130.23, 129.97, 128.34, 128.29, 128.12, 127.23, 116.74, 113.39, 86.55, 85.47, 73.80, 73.73, 63.99, 55.43, 42.47, 36.81.

4.7. 2'-Deoxy-5'-dimethoxytrityl-1-methyl-N-benzoylpseudocytidine 3'-(2-cyanoethyl diisopropylphosphoramidite) (13)

To a mixture of 2'-deoxy-5'-dimethoxytrityl-1-methyl-N-benzoyl-pseudocytidine (12) (120 mg, 0.185 mmol) and DIPEA (97 µL, 0.55 mmol) in anhydrous CH₂Cl₂ (5 mL) was added 2-cyanoethyl diisopropylchlorophosphoramidite (62 µL, 0.278 mmol) at 0 °C. The mixture was allowed to warm to rt and stirred for 2 h. The mixture was poured into a saturated sodium bicarbonate solution (10 mL) and extracted with CH2Cl2. The organic layer was washed with brine, dried over anhydrous sodium sulfate and evaporated. The product was purified from the residue by flash chromatography (neutral silica, EtOAc/hexane = 1:3 to 1:1 with 0.5% Et₃N) to give diastereomeric mixture of 13 as a white foam (123 mg, 78%). TLC (EtOAc/hexane, 1:1): Rf 0.5, 0.53. ¹H NMR (300 MHz, CDCl3) & 8.2-8.3 (m, 2H), 7.2-7.8 (m, 14H), 6.8-6.9 (m, 4H), 5.39 (m, 1H), 4.57 (m, 1H), 4.19 (m, 1H), 3.79 (m, 6H), 3.5-3.7 (m, 3H), 3.2-3.5 (m, 2H), 3.15 (m, 3H), 2.7-2.9 (m, 1H), 2.4-2.6 (m, 2H), 2.0 (m, 1H), 1.6 (m, 1H), 1.0-1.3 (m, 12H). ³¹P NMR (121 MHz, CDCl₃) δ 150.26 (s), 149.36 (s).

4.8. Oligonucleotide synthesis

Oligonucleotide synthesis was performed by the DNA synthesizer ABI 394, at the 1.0 µmol scale. The synthesis was done using the manufacturer's recommended procedure except extended (600 s) coupling time for isoG, isoC and ψ C. All natural (A, G, C, and T) and unnatural (isoG and isoC) nucleobase phosphoramidites were purchased from Glen Research. At the completion of the synthesis, all protective groups in oligonucleotides were removed by heating in 30% ammonium hydroxide at 55 °C overnight. The mixture was then filtered through 0.2 µm filter and the remaining ammonia was removed by rotary evaporation under reduced pressure at rt. The sample was then diluted with water and lyophilized. The residue after lyophilization was dissolved in water and filtered through 0.2 µm filter. The oligonucleotide was purified from the filtrate by ion exchange HPLC (Dionex DNAPac PA-100 9 × 250 mm Semi-Prep column, eluent A = 25 mM NaOH, eluent B = 25 mM NaOH, 1.0 M NaCl, gradient from 10% to 80% B in 40 min, flow rate 3 mL/min). The appropriate fractions were collected, neutralized by aqueous acetic acid, desalted by Sep-PaK cartridges (Waters) and lyophilized. The purified oligonucleotides were dissolved in water at a concentration of 50 µM.

4.9. HPLC profile of synthetic oligonucleotides

HPLC condition; Dionex DNAPac PA-100 4 × 250 mm analytical column, eluent A = 25 mM NaOH, eluent B = 25 mM NaOH, 1.0 M NaCl, gradient from 10% to 80% B in 50 min, flow rate 0.5 mL/min 5'-CACNIACTTTCTCCT-3'

3'-TG T G N2 T G A A AG A G G-5'

N ₁	Retention time N ₂	Retention time
iG	36.6 <i>i</i> G	43.5
iC	34.4 iC	41.8
ψC	34.3 ΨC	41.8

4.10. Thermal DNA duplex denaturation

Thermal DNA duplex denaturation experiments were done in 0.45 M NaCl, 0.045 M sodium citrate buffer (pH 7.9). Each oligonucleotide was present at 1.6 µM. Absorbance was monitored at 260 nm over a range of 25 °C to 85 °C with a change in temperature of 1 °C/min⁻¹ for five heating and cooling cycles on a Varian Cary 300 (Palo Alto, CA). Initial heating and cooling cycles were discarded and the $T_{\rm m}$ was determined by averaging the temperature of the maximum derivatives of each cooling and heating cycle.

4.11. Acid stability

The 5'-radiolabelled oligonucleotides (10 µL, 0.2 pmol/µL) were mixed with an equal volume of 0.1 M acetic acid. Mixtures were incubated for 0, 15, 30, 60, or 90 min at 95 °C in a thermocycler with a heated lid. The tubes containing the mixtures were then heated with the tubes open to let volatile materials evaporate (2 min). Two volumes (20 µL) of ammonium hydroxide (0.1 M) were then added, and heating continued at 95 °C for 5 min. The tubes were then opened, and the NH₃ was evaporated by heating at 95 °C for 2 min. Cleavage products were resuspended in formamide (3×) and resolved by PAGE (20%, 0.3 pmol samples).

4.12. Thermal stability

To determine stability to heating at near neutral pH, oligonucleotides (10 µL, 0.2 pmol/µL) in Tris buffer (10 mM, pH 8.5) were incubated for 0, 15, 30, 60, or 90 min at 95 °C in a thermocycler with a heated lid. The tubes containing the mixtures were then heated with the tubes open to let volatiles evaporate (2 min). Cleavage products were resuspended in formamide (3x) and resolved by PAGE.

Acknowledgement

This work was supported by the National Institutes of Health 1R01GM081527-01 and a grant from SAB.

References and notes

- 1. Sismour, A. M.: Benner, S. A. Nat. Rev. Genet. 2005, 6, 533.
- Rich, A. In Horizons in Biochemistry; Kasha, M., Pullman, B., Eds.; Academic Press: NY, 1962; p 103.
- Elbeik, T.; Markowitz, N.; Nassos, P.; Kumar, U.; Beringer, S.; Haller, B.; Ng, V. J. Clin. Microbiol. 2004, 42, 3120.
- 4. Benner, S. A. Acc. Chem. Res. 2004, 37, 784.
- Sismour, A. M.; Benner, S. A. Nucleic Acids Res. 2005, 33, 5640. 6. Johnson, S. C.; Sherrill, C. B.; Marshall, D. J.; Moser, M. J.; Prudent, J. R. Nucleic
- Acids Res. 2004, 32, 1937. 7. Hutter, D.; Benner, S. A. J. Org. Chem. 2003, 68, 9839.
- 8.
- Havemann, S. A.; Hoshika, S.; Hutter, D.; Benner, S. A. Nucleosides Nucleotides Nucleic Acids 2008, 27, 261. 9. Bhattacharya, B. K.; Devivar, R. V.; Revankar, G. R. Nucleosides Nucleotides 1995,
- 14, 1269, 10. Zhang, H.-C.; Daves, D. D., Jr. J. Org. Chem. 1992, 57, 4690.
- 11. Geyer, C. R.; Battersby, T. R.; Benner, S. A. Structure 2003, 11, 1485.
- 12. Horn, T.; Chang, C. A.; Collins, M. L. Tetrahedron Lett. 1995, 36, 2033.
- 13. Seela, F.; He, Y. J. Org. Chem. 2003, 68, 367.